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Chapter 19 Study Guide Acids

An acid is a substance that contains hydrogen and ionizes to produce hydrogen ions in aqueous solution; A base is a substance that contains a hydroxide group and dissociates to produce a hydroxide ion in aqueous solution.

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Acids: Compounds that yield hydrogen ions (H^+) in an aqueous s.... monoprotic acids - 1 H^+ ... diprotic acids - 2 H^+ ... triprotic acids.... Properties of Acid. Taste sour, turns litmus paper red, conducts electricity. Properties of Bases. Tastes bitter, feels slippery, turns litmus paper blue, conduc.... 27 terms.

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The Acids, Bases and Salts chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with acids, bases and salts.

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Created by. harvestjoy26. Terms in this set (36) Properties of Acids. Conducts electricity, tastes sour, electrolytes, neutralizes bases, 1-6, affects indicators. Properties of Bases. Conducts electricity, tastes bitter, feels slippery, electrolytes, neutralizes acids, 8-14, affects indicators. Arrhenius acids. Yields H^+ ions.

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19 Date Class Section Strengths of Acids and Bases In your textbook, read about strengths of acids. Circle the letter of the choice that best completes the statement or answers the question. I. Acid A and acid B are of equal concentration and are tested with a conductivity apparatus. When the electrodes are placed in acid A, the bulb glows dimly.

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KEY Study Guide: Acids and Bases 1) List at least three
characteristic properties of acids and three of bases. ACIDS pH
less than 7 Have a sour taste Change the color of many
indicators Are corrosive (react with metals) Neutralize bases
Conduct an electric current

KEY - acid base study guide

Study Guide with Selected Solutions for Stoker's General,
Organic, and Biological Chemistry (6th Edition) Edit edition.
Problem 29EP from Chapter 10: Classify each of the acids in
Problem 10.19 as a strong acid...

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