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Chapter 11: Chemical Reactions Study Guide. Lily Taylor. 19 October 2020 . question. chemical equation. answer. A representation of a chemical reaction with reactants on the left, products on the right, and an arrow separating the two. question. skeleton equation. answer. A chemical equation that does not indicate the relative amounts of the ...

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Section 11.1 Assessment. Describe the steps in writing a balanced chemical equation. Write the skeleton equation for the

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following reactions: Heating solid copper(II)sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride. $\text{CuS (s)} + \text{O}_2\text{(g)} \rightarrow \text{Cu (s)} + \text{SO}_2\text{(g)}$

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Demitri_Bautista. Key Concepts: Terms in this set (253) chemical equation _____ is a representation of a chemical reaction with reactants on the left, products on the right, and an arrow separating the two.

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322 Chapter 11 Figure 11.2 Three common chemical reactions are shown. When methane gas burns, it combines with oxygen to form carbon dioxide and water. Iron turns to red-brown rust (iron(III) oxide) in the presence of oxygen in the air. Hydrogen peroxide decomposes to water and oxygen when used as an antiseptic. a b c

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11.1 Describing Chemical Reactions 11

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As shown in Figure 11.3.1, applying a small amount of heat to a pile of orange ammonium dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano. Heat, light, and gas are produced as a large pile of fluffy green chromium (III) oxide forms. We can describe this reaction with a chemical equation An expression that gives the identities and quantities of the substances in a chemical reaction.

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